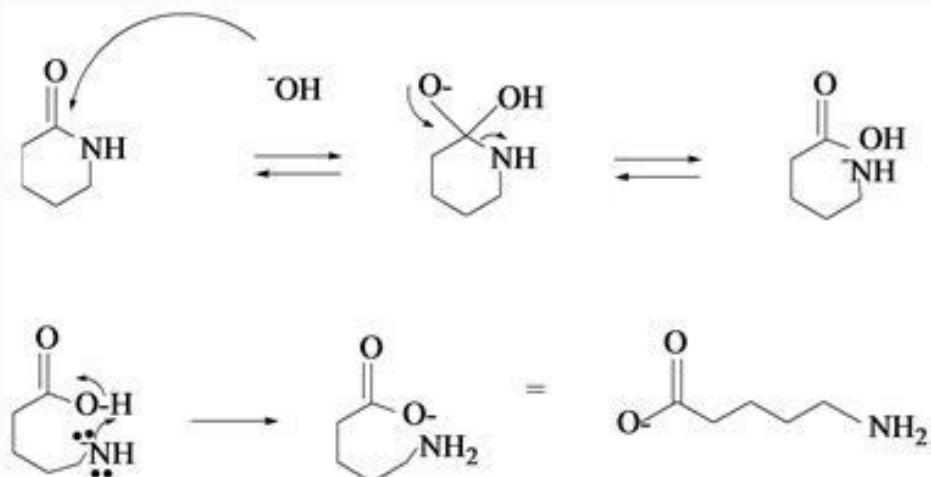
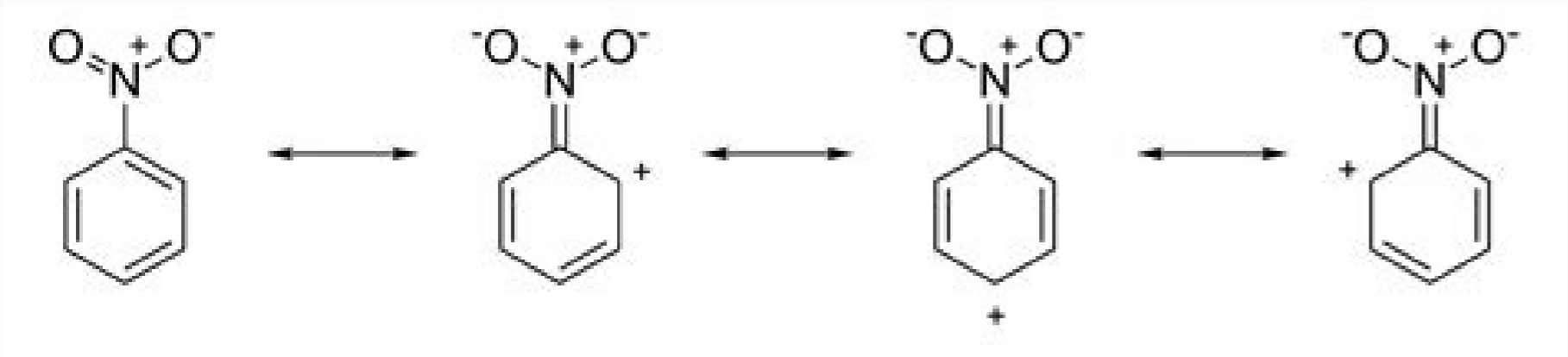
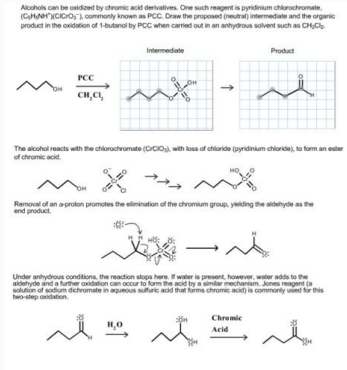
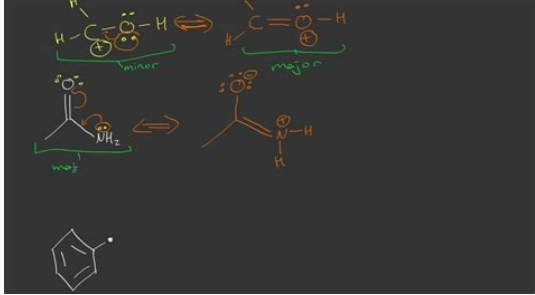


I'm not robot!



Formula of organic compounds. Formula of organic chemistry. Concept of organic chemistry.

It has been noted in the comments and I shall repeat that the premise of the question is already bogus. The question does not consider two different ions but two mesomeric depiction of the same ion. This is also evident by the mesomeric arrow between the two forms. Since mesomerism strictly means that no geometrical change takes place, both depictions are identical in all aspects and have the same (resonance-influenced) energy. Thus, any book claiming that either of the two have a lower energy is wrong. However, what is commonly done is to determine a more likely resonance structure. This is, for example, commonly used to explain the strongly stabilised nature of certain aromatic systems, e.g. when DMAP (4-*N,N*-dimethylaminopyridine) is used as a carboxylic activating agent. We can attempt to do that for your structure, too. If we compare the two we see — of course — that a lot of features are identical. The principal differences are: A has a trisubstituted double bond while B only as a disubstituted (and terminal) double bond B has a secondary carbocation while A has a primary one (both, however, are allylic). Since allylic stabilisation for cations is much stronger than hyperconjugation, we can discard the second bullet point as being essentially equal for both resonance structures. Thus, we should argue with the location of the double bond. Both double bonds can benefit from hyperconjugation twice (the primary carbon of A cannot hyperconjugate due to geometry), but in A the two hyperconjugations stabilise different carbons. Along with the general double bond stability given as: $\text{tetrasubstituted} < \text{trisubstituted} < \text{disubstituted} < \text{monosubstituted} < \text{terminal disubstituted} < \text{cis} < \text{trans} < \text{trisubstituted} < \text{tetrasubstituted}$ we can identify A as being slightly better than B and therefore contributing slightly more towards the overall structure. You mentioned in the comments that your book states: structure B is not planar and hence conjugation is not possible in it. Your book probably wants you to assume that the following depiction in figure 1 is adequate for the two resonance forms. Figure 1: structures B and A under the assumption of a chair-configured cyclohexyl structure for B. Under the underlying assumption of a different geometry for both structures, this may be valid. However, there is no way that cation B will adopt a chair cyclohexane configuration. Instead, as I mentioned above, the following structures in figure 2 are much more likely and even show that we are dealing with resonance. Figure 2: depiction of the mesomeric forms without geometry change. You can easily see that the argument of non-conjugation is bogus, since there very well is a way to optimise geometry so that the orbitals may form an allyl system. canonical form (plural canonical forms) (mathematics, computing) A standard or normal presentation of a mathematical entity. (chemistry) Any of a set of representations of the resonance structure of a molecule each of which contributes to the real structure; a contributing structure. What is canonical structure? A molecule or ion with such delocalized electrons is represented by several contributing structures (also called resonance structures or canonical structures). What is canonical in SEO? The rel=canonical element, often called the "canonical link", is an HTML element that helps webmasters prevent duplicate content issues. It does this by specifying the "canonical URL", the "preferred" version of a web page — the original source, even. Using it well improves a site's SEO. The rel=canonical element, often called the "canonical link", is an HTML element that helps webmasters prevent duplicate content issues. It does this by specifying the "canonical URL", the "preferred" version of a web page — the original source, even. Using it well improves a site's SEO. A canonical model is a design pattern used to communicate between different data formats. A form of enterprise application integration, it is intended to reduce costs and standardize on agreed data definitions associated with integrating business systems. A canonical link element is an HTML element that helps webmasters prevent duplicate content issues by specifying the "canonical" or "preferred" version of a web page as part of search engine optimization. It is described in RFC 6596, which went live in April 2012. The 20 amino acids that are encoded directly by the codons of the universal genetic code are called standard or canonical amino acids. A modified form of methionine (N-formylmethionine) is often incorporated in place of methionine as the initial amino acid of proteins in bacteria, mitochondria and chloroplasts. A canonical tag (aka "rel canonical") is a way of telling search engines that a specific URL represents the master copy of a page. Using the canonical tag prevents problems caused by identical or "duplicate" content appearing on multiple URLs. (adj.) Authoritative or standard; conforming to an accepted rule or procedure. When referring to programming, canonical means conforming to well-established patterns or rules. The term is typically used to describe whether or not a programming interface follows the already established standard. It is also used in mathematics, music and can refer to something reduced to its most basic form. The word canonical is from the root canon, with both evolving from the Latin cononicus, or "according to rule," a meaning applied to religion during the Middle Ages. The Christian Church's recognition of books of the Old Testament and the New Testament and of the four gospels- the four "canonical gospels", (Matthew, Mark, Luke, and John) occurred gradually during the first four centuries. A canonical name is the properly denoted host name of a computer or network server. A CNAME specifies an alias or nickname for a canonical host name record in a domain name system (DNS) database. In programming, the term "canonical" means "according to the rules." In molecular biology and bioinformatics, the consensus sequence (or canonical sequence) is the calculated order of most frequent residues, either nucleotide or amino acid, found at each position in a sequence alignment. Such information is important when considering sequence-dependent enzymes such as RNA polymerase. Resonance structures are two forms of a molecule where the chemical connectivity is the same but the electrons are distributed differently around the structure. Resonance occurs when electrons can flow through neighboring pi systems. Canonical Ltd. is a UK-based privately held computer software company founded and funded by South African entrepreneur Mark Shuttleworth to market commercial support and related services for Ubuntu Linux and related projects. In chemistry, resonance or mesomerism is a way of describing delocalized electrons within certain molecules or polyatomic ions where the bonding cannot be expressed by one single Lewis structure. From an SEO point of view here is the definition of a canonical URL. Canonical URL: the search engine friendly URL that you want the search engines to treat as authoritative. In other words, a canonical URL is the URL that you want visitors to see. Quite often canonical URLs were used to describe the homepage. Benzene has two resonance structures, showing the placements of the bonds. Another example of resonance is ozone. Ozone is represented by two different Lewis structures. The difference between the two structures is the location of double bond. noncanonical. [non-kuh-non-i-kuh] See more synonyms on Thesaurus.com adjective. not included within a canon or group of rules. not belonging to the canon of Scripture. The resonance effect is a chemical phenomenon observed in compounds characteristic of double bonds of organic compounds. Organic compounds that contain double bonds in their structure are usually made of the overlap of p-orbitals on two adjacent carbon atoms (referred to as pi bonds). Hyperconjugation is the stabilising interaction that results from the interaction of the electrons in a sigma bond (usually C-H or C-C) with an adjacent empty or partially filled p-orbital or a n-orbital to give an extended molecular orbital that increases the stability of the system. Pushing a person in a swing is a common example of resonance. The loaded swing, a pendulum, has a natural frequency of oscillation, its resonant frequency, and resists being pushed at a faster or slower rate. Resonance is a mental exercise and method within the Valence Bond Theory of bonding that describes the delocalization of electrons within molecules. 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